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Ring-closing olefin metathesis reactions; synthesis of iso-β-bisabolol

Jutta M. Mörgenthaler and Dietrich Spitzner*

Institut für Chemie, Abt. Bioorganische Chemie, Universität Hohenheim, Garbenstraße 30, D-70599 Stuttgart, Germany

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Abstract—A ring-closing olefin metathesis is the key step in a synthesis of the rare iso- β -bisabolol found in sandalwood oils. © 2003 Elsevier Ltd. All rights reserved.

The bisabolols were isolated from *Santalum* spp. Besides the odourless major bisabolols **2** and **3**, trace constituents possessing the isomeric structures **1** and **4** have been found very recently and synthesized and evaluated for their odour impact. **1** is an interesting odour active compound with a strong floral scent.¹ Herein we report a straightforward synthesis of a mixture of the four stereoisomers of bisabolol **1** using a ring-closing olefin metathesis (RCM) reaction as the key step.² dized with PCC to ketone 7. The tertiary alcohol 5 was obtained from ketone 7 by the addition of in situ generated allyllithium 6 (from allyltriphenyltin and phenyllithium).^{8,9} Selective ring-closing olefin metathesis was achieved using Grubbs second generation catalyst 11 to yield iso- β -bisabolol 1 as a mixture of all four stereoisomers. This mixture had been separated previously by the DRAGOCO group.¹ The number of stereoisomers of 1 can be reduced by using enantiomerically pure 10.⁷ This synthetic scheme will be applied to the synthesis of



A retrosynthetic analysis (Scheme 1) leads to tri-olefin 5, which should be regioselectively cyclized by an olefin metathesis reaction.

The required tri-olefin **5** was prepared by the sequence described in Scheme 2.³ Conjugate addition⁴ of the Grignard reagent **8** to methyl methacrylate in the presence of CuBr–Me₂S complex and TMSCl yielded an ester, which was reduced to the known alcohol **9**.^{5,6} Oxidation of **9** with PCC gave the unstable aldehyde **10**,⁷ which was reacted with the Grignard reagent **8** to give an intermediate secondary alcohol, which was oxi-

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2 starting from aldehyde **10a**.^{10,11} Fráter and Müller had previously prepared bisabolol **2** by a different route.¹²



Scheme 1.

^{*} Corresponding author. Tel.: +49-711-459-2812; fax: +49-711-459-3703; e-mail: spitzner@uni-hohenheim.de

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Scheme 2. Reagents and conditions: (a) CuBr–Me₂S, LiI, TMSCl, Ar, -78 °C, 4 h, 53%; (b) LAH, ether, 2 h, 91%; (c) PCC, CH₂Cl₂, rt, 2 h, 89%; (d) 8, THF, reflux, 20 min, 55%; (e) PCC, CH₂Cl₂, rt, 2 h, 86%; (f) 6, ether, reflux, Ar, 80%; (g) 11, benzene, 80 °C, Ar, 4 h, 93%.

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